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## Structure Reports

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## Bis(triphenylphosphine)iminium tricarbonylcyclopentadienylmolybdate

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## Key indicators

Single-crystal X-ray study
$T=203 \mathrm{~K}$
Mean $\sigma(\mathrm{C}-\mathrm{C})=0.004 \AA$
$R$ factor $=0.037$
$w R$ factor $=0.090$
Data-to-parameter ratio $=17.8$

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.
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The crystal structure of bis(triphenylphosphine)iminium cyclopentadienyltricarbonylmolybdate, $[\mathrm{PPN}]\left[\mathrm{Mo}(\mathrm{CO})_{3}\left(\eta^{5}-\right.\right.$ $\left.\left.\mathrm{C}_{5} \mathrm{H}_{5}\right)\right]\left(\mathrm{PPN}^{+}=\left[\left(\mathrm{PPh}_{3}\right)_{2} \mathrm{~N}\right]^{+}, \mathrm{C}_{36} \mathrm{H}_{30} \mathrm{NP}_{2}\right)$, is reported. The anion is configured in the 'piano-stool' arrangement while the cation adopts a bent configuration about the $\mathrm{P}-\mathrm{N}-\mathrm{P}$ bond. An 'expanded-phenyl-embrace' supramolecular motif is noted in the packing of the $\mathrm{PPN}^{+}$cation.

## Comment

Though a commonly used metal carbonyl reagent, the structure of $[\mathrm{PPN}]\left[\mathrm{Mo}(\mathrm{CO})_{3}\left(\eta^{5}-\mathrm{C}_{5} \mathrm{H}_{5}\right)\right]$, (I), has not been previously reported.

(I)

The anion adopts the classic piano-stool-type arrangement common for $\mathrm{Cp} M L_{3}$ systems. The average $\mathrm{Mo}-\mathrm{C}_{\mathrm{CO}}$ bond length $[1.930$ (3) $\AA$ A is slightly larger than that reported for the tetrabutylammonium salt $[1.909$ (9) $\AA$ ] while the average $\mathrm{C}_{\mathrm{CO}}-\mathrm{Mo}-\mathrm{C}_{\mathrm{CO}}$ angle $\left[87(2)^{\circ}\right]$ and $\mathrm{Mo}-\mathrm{C}_{\mathrm{Cp}}$ distance [2.37 (2) A] are similar to those reported [88.1(3) and 2.37 (1) A respectively; Crotty et al., 1977]. The PPN ${ }^{+}$cation adopts the typical bent configuration about the $\mathrm{P}-\mathrm{N}-\mathrm{P}$ bond [142.62 (14) ${ }^{\circ}$ ].

A characteristic of triphenylphosphine-related systems is the observance of 'phenyl embraces' as a crystal packing motif (Scudder \& Dance, 1998). These embraces involve intermolecular phenyl attractions (both edge-face and offset faceface attractions) forming extended networks through the crystal lattice. This type of packing motif is noted in the structure of $[\mathrm{PPN}]\left[\mathrm{Mo}(\mathrm{CO})_{3}\left(\eta^{5}-\mathrm{C}_{5} \mathrm{H}_{5}\right)\right]$. The non-bonded $\mathrm{P}-$ P distance of 6.8606 (8) $\AA$ and $\mathrm{N}-\mathrm{P}-\mathrm{P}$ angle of 72.41 (7) ${ }^{\circ}$ are characteristic of an 'expanded-phenyl-embrace' crystal packing motif in a compound containing $\mathrm{PPN}^{+}$(Lewis \& Dance, 2000).

## Experimental

$\left[\mathrm{Mo}(\mathrm{CO})_{3}\left(\eta^{5}-\mathrm{C}_{5} \mathrm{H}_{5}\right)\right]_{2}(0.2 \mathrm{~g}, 0.41 \mathrm{mmol})$ was reduced in tetrahydrofuran ( 20 ml ) over a $1 \% \mathrm{Na} / \mathrm{Hg}$ amalgam for $2-3 \mathrm{~h}$. A dichloromethane solution ( 5 ml ) of $[\mathrm{PPN}] \mathrm{Cl}(0.47 \mathrm{~g}, 0.82 \mathrm{mmol})$ was added and the reaction mixture stirred for an additional 30 min . The pale

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yellow-orange solution was transferred by syringe and solvent removed under vacuum. Recrystallization by addition of diethyl ether to a dichloromethane solution of the title compound produced a pale yellow powder. Crystals suitable for crystallographic analysis were obtained by liquid-liquid diffusion under a nitrogen atmosphere of diethyl ether and light petroleum spirits into a dichloromethane solution of the compound

## Crystal data

$\mathrm{C}_{36} \mathrm{H}_{30} \mathrm{NP}_{2} \cdot \mathrm{C}_{8} \mathrm{H}_{5} \mathrm{MoO}_{3}$
$M_{r}=783.61$
Monoclinic, $P 2_{1} / c$
$a=14.0385$ (2) $\AA$
$b=19.1269$ (2) $\AA$
$c=13.8322$ (1) $\AA$
$\beta=90.728(1)^{\circ}$
$V=3713.83(7) \AA^{3}$
$Z=4$

## Data collection

Siemens SMART CCD
diffractometer
Multiscan scans
Absorption correction: empirical (Blessing, 1995)
$T_{\text {min }}=0.885, T_{\text {max }}=0.910$
22556 measured reflections

## Refinement

Refinement on $F^{2}$
$R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.037$
$w R\left(F^{2}\right)=0.090$
$S=1.08$
8177 reflections
460 parameters
H -atom parameters constrained
H atoms were placed in calculated positions, with $U_{\text {iso }}$ values 1.2 times the $U_{\text {iso }}$ values of the atoms to which they are attached.

Data collection: SMART (Siemens, 1994); cell refinement: SAINT (Siemens, 1994); data reduction: $S A D A B S$ (Blessing, 1995); program(s) used to solve structure: SHELXS-97 (Sheldrick, 1997);


Figure 1
Structure of $[\mathrm{PPN}]\left[\mathrm{Mo}(\mathrm{CO})_{3}\left(\eta^{5}-\mathrm{C}_{5} \mathrm{H}_{5}\right)\right]$ showing the atom-labelling scheme. Ellipsoids are drawn at the $50 \%$ probability level (Farrugia, 1997)
program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: ORTEP-3 for Windows (Farrugia, 1997); software used to prepare material for publication: WinGX (Farrugia, 1999).

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## References

Blessing, R. H. (1995). Acta Cryst. A51, pp. 33-38.
Crotty, D. E., Corey, E. R., Anderson, T. J., Glick, M. D. \& Oliver, J. P. (1977). Inorg. Chem. 16, 920-924.
Farrugia, L. J. (1997). J. Appl. Cryst. 30, 565.
Farrugia, L. J. (1999). J. Appl. Cryst. 32, 837-838.
Lewis G. R. \& Dance, I. (2000). J. Chem. Soc. Dalton Trans. pp. 299-306.
Scudder, M, \& Dance, I. (1998). J. Chem. Soc. Dalton Trans. pp. 329-344.
Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. Universität Göttingen, Germany.
Siemens (1994). SMART, SAINT and SHELXTL. Siemens Analytical Instruments Inc., Madison, Wisconsin, USA.

